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Jones reagent oxidizes a secondary alcohol (such as 2-propanol) to the corresponding ketone (acetone). Jones reagent is aqueous and strongly acidic.

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Jones Oxidation - Organic Chemistry

Jones Reagent. The Jones Reagent is a solution of chromium trioxide in diluted sulfuric acid that can be used safely for oxidations of organic substrates in acetone. The reagent can also be prepared from sodium dichromate and potassium dichromate. Jones Reagent is especially suitable for the oxidation of secondary alcohols to ketones and of primary

Solutions Manual of Organic Chemistry by Jones & Gingrich

Maitland Jones, Jr., is Professor of Chemistry at New York University. His undergraduate and graduate degrees are from Yale University. A long-time teacher of the organic chemistry course at Princeton, he is also an internationally recognized research scientist.

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The Jones Oxidation allows a relatively inexpensive conversion of secondary alcohols to ketones and of most primary alcohols to carboxylic acids. The oxidation of primary allylic and benzylic alcohols gives aldehydes. Jones described for the first time a conveniently and safe procedure for a chromium (VI)-based oxidation, that paved the way for

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